

New synthesis of quinolines from lithiated allenes and isothiocyanates

L. Brandsma,^a N. A. Nedolya,^{b*} R.-J. de Lang,^a and B. A. Trofimov^b^aPreparative Organic Chemistry Department, Utrecht University,
Padualaan 8, 3584 CH, Utrecht, The Netherlands^bIrkutsk Institute of Organic Chemistry, Siberian Branch of the Russian Academy of Sciences,
1 ul. Favorskogo, 664033 Irkutsk, Russian Federation.
Fax: 007 (395 2) 35 6046

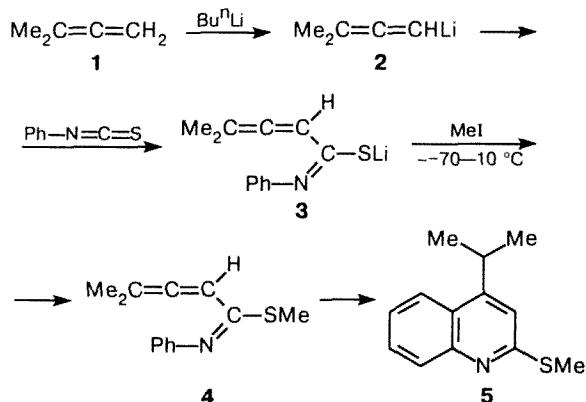
A new route was proposed for the synthesis of quinolines based on the reaction of lithiated allenes with phenyl isothiocyanate.

Key words: phenyl isothiocyanate, allenes, butyllithium, lithiation, alkylation, azatrienes, quinolines.

Data on the synthesis of quinolines by the reaction of organometallic compounds with isothiocyanates is lacking in the literature.¹

We carried out for the first time the synthesis of quinolines, 4-isopropyl-2-methylthioquinoline (**5**) being an example, (Scheme 1) by adding lithium 3,3-dimethylallenide (**2**) to phenyl isothiocyanate followed by alkylation of thiolate **3** with methyl iodide to yield 1-aza-1,3,4-triene (**4**). Compound **4** was isolated in a quantitative yield. It transformed into quinoline **5** in an exothermal reaction, which started already at -40°C . The structure of compound **5** was confirmed by the data from IR and ^1H and ^{13}C NMR spectroscopy.

Scheme 1



Experimental

The IR spectra were recorded on a Specord IR-75 spectrophotometer in thin films. The NMR spectra were recorded on

Varian EM-390 (90 MHz; $\sim 20\%$ solution in CCl_4 ; Me_4Si as the internal standard) and Bruker AC-300 (300 and 75 MHz for ^1H and ^{13}C NMR spectra, respectively; $\sim 20\%$ solution in CDCl_3 ; Me_4Si as the internal standard) spectrometers. A GLC analysis was carried out on a Varian 3400 gas chromatograph (flame ionization detector; $15\text{ m} \times 0.53\text{ mm}$ capillary column; the thickness of the DB-5 liquid phase was $1.5\text{ }\mu\text{m}$; nitrogen was the carrier gas).

All operations were carried out in an atmosphere of nitrogen. 1,1-Dimethylallene (**1**) was synthesized according to the procedure reported in Ref. 2. Tetrahydrofuran was purified with mechanically dispersed KOH ($\sim 50\text{ g L}^{-1}$) and by distillation over LiAlH_4 in the presence of benzophenone in an atmosphere of nitrogen. Butyllithium (1.6 M solution in hexane) was supplied by Chemetall (Germany).

4-Isopropyl-2-methylthioquinoline (5). 1,1-Dimethylallene (**1**) (9.8 g, 0.14 mol) was added to a solution of Bu^nLi (0.11 mol) in dry THF (80 mL) and hexane (70 mL) cooled to -60°C . The mixture was stirred at -30°C for 50 min. Then the mixture was cooled to -100°C , and a solution of phenyl isothiocyanate (13.5 g, 0.10 mol) in THF ($\sim 20\text{ mL}$) was added portionwise (the temperature was maintained at -100°C). The reaction mixture was kept for 15 min, during which its temperature increased to -70°C . Then methyl iodide (19 g, 0.13 mol) was added. The mixture was stirred at 10°C for 10–15 min and then diluted with cold water ($\sim 60\text{ mL}$). The organic layer was separated. The aqueous layer was extracted with pentane and ether. The combined organic layers were dried with potassium carbonate, and the solvent was evaporated. Azatriene (**4**) was obtained in a yield of 21.7 g (100%), n_{D}^{20} 1.6054. ^1H NMR (90 MHz, CCl_4), δ : 6.65–7.30 (m, 5 H, Ph); 5.65 (m, 1 H, $\text{CH}=\text{C}$); 2.32 (s, 3 H, SMe); 1.76 (s, 6 H, 2 Me).

After heating of compound **4** to -40°C , the exothermal heterocyclization started (the temperature increased spontaneously to -245°C). Distillation *in vacuo* gave 4-isopropyl-2-methylthioquinoline (**5**) in a yield of 16 g (74%), the purity was 96% (GLC), b.p. $125\text{--}145^{\circ}\text{C}$ ($\sim 0.5\text{ Torr}$), n_{D}^{20} 1.6285. Found (%): C, 71.46; H, 6.93; N, 6.48; S, 15.24. $\text{C}_{13}\text{H}_{15}\text{NS}$. Calculated (%): C, 71.89; H, 6.91; N, 6.45; S, 14.75. IR, v/cm^{-1} : 600 w, 680 w, 750, 830 v.w, 860, 900, 960 w, 1020 w,

1070, 1100, 1140, 1160, 1290, 1375 w, 1400, 1450 w, 1490 v.w, 1540, 1590, 2860, 2920, 2960, 3050. ^1H NMR (300 MHz, CDCl_3), δ : 8.11 (dd, 1 H, 8-H); 7.95 (dd, 1 H, 5-H); 7.65 (td, 1 H, 7-H); 7.44 (td, 1 H, 6-H); 7.24 (s, 1 H, 3-H); 3.58 (m, 1 H, CH); 2.79 (s, 3 H, SMe); 1.36 (d, 6 H, 2 Me). ^{13}C NMR, δ : 159.31 (2-C); 152.70 (8a-C); 148.37 (4-C); 128.57, 124.46, 122.79 (5-C—8-C); 124.40 (4a-C); 116.04 (3-C); 27.82 (CH); 22.31 (Me).

References

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